M.Sc. 4th Semester Examination, 2021 CHEMISTRY

(Inorganic Chemistry Special)

Paper: CHEM 401E

Course ID: 41451

Time: 2 Hours Full Marks: 40

The figures in the right-hand side margin indicate full marks.

Candidates are required to give their answers in their own words

as far as practicable

1. Answer *any five* of the following questions:

 $2 \times 5 = 10$

- (a) Give any two essential characteristics which a nuclide has for exhibiting the Mössbauer effect.
- (b) What case will result in Mössbauer spectrum for a delta shift of 0.00 mm/s?
- (c) Give one example of molecular rectifiers.
- (d) What causes white rust?
- (e) How do you remove the tosyl group from N-protected Azathiacrowns?
- (f) "Choice of the base is crucial to the successful synthesis of crown ethers"- Explain.
- (g) What is atmospheric corrosion?
- 2. Answer *any four* of the following questions:

 $5 \times 4 = 20$

- (a) (i) Arrange the isomer shift value of the following ions in increasing order: Sn, Sn^{2+} and Sn^{4+} . Give suitable explanation.
 - (ii) How many numbers of allowed transitions are possible in the Mössbauer spectrum of an iron containing sample, recorded in the presence of a static magnetic field.

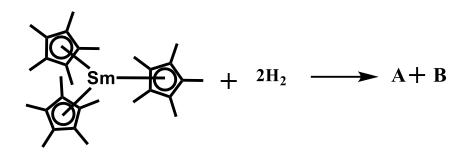
3+2=5

(b) (i) In the ⁵⁷Fe Mossbauer experiment source of 25.5 keV is moved towards the absorber at a velocity of 5.5 mms⁻¹. Calculate the shift in frequency of the source for this sample in MHz.

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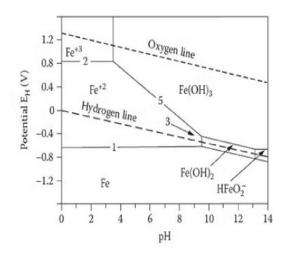
- (ii) Which one will be corroded at a higher rate in deaerated HCl medium among iron and zinc and why?

 3+2=5
- (c) (i) Why caesium carbonate is preferred over caesium compounds in macrocyclization reactions as basic reagents?
 - (ii) What is the difference between Carcerands and Hemicarcerands? Give one application of hemicarceplexes. 2+(1+2)=5
- (d) (i) Give one example of water-soluble cryptophanes along with its applications.
 - (ii) What is the role of macrocycles in Phase-transfer catalysis (PTC)? 3+2=5
- (e) (i) Write down the structure of A and B.



- (ii) Comment on the formulae of the alkyls formed by the lanthanides with the CH₃, CH₂SiMe₃, CMe₃ and CH(SiH₃)₂ ligands.
- (iii) Predict the geometry of [Ln(CH₂SiMe₃)₃(THF)₂]. 2+2+1=5
- (f) Give one example of each of the NAND and OR logic gates. 2.5+2.5=5
- 3. Answer *any one* of the followings: $10 \times 1 = 10$
 - (a) (i) The compound $K_4[Fe(CN)_6]\cdot 3H_2O$ gives single line Mössbauer spectrum with no quadrupole splitting-Explain
 - (ii) Calculate the recoil velocity and energy of a Mossbaur nucleus having at.wt. 60 if the emitted gamma-ray has a frequency 1.84×1014 Hz. ($N = 6.023 \times 1023$, $h = 6.626 \times 10-34$ Js, $c = 3.0 \times 108$ m.s -1)
 - (iii) Explain the following terms in the instrumentation of Mossbauer spectroscopy: vibrator and absorber.
 - (iv) What do you mean by "template effect"? 2+4+2+2=10

(b) Pourbaix diagram for the iron-water system shown below



- (i) Using the Pourbaix diagram for Fe provided, write a balanced half-reaction for reduction of Fe (III) to Fe (II) at pH = 5.
- (ii) Calculate the potential at pH= 7, considering the Fe(III)/Fe(II) reduction potential +0.45 V at pH = 5.
- (iii) Why the line to the right of Fe³⁺ in the Pourbaix diagram is vertical?
- (iv) How can old iron objects survive a few hundred years under water? 3+3+2+2=5
